	[Title]		[Instructor]		
	A	dvanced Physical Chemistry	Kenji Miyatake / Shinji Nohara / Akiyoshi Kuzume		
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG531	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester Fri./I		English/ Japanese

Physical chemistry is one of the most important basic subjects in the field of green energy conversion technology. In this class, basics of physical chemistry will be studied in preparation for the advanced classes.

[Objectives]

To understand physical properties of gases, thermodynamics, phase equilibrium, electrochemistry, and reaction kinetics

[Requirements]

Basic knowledge on physical chemistry

[Evaluation]

Report and examination: 60%

Mark given for class participation: 40%

[Textbooks]

None

[References]

Physical Chemistry (Gordon M. Barrow), ISBN-10: 0070051119

- 1. The properties of gases
- 2. Thermodynamics and the first law
- 3. Thermochemistry
- 4. Entropy, the second and the third law
- 5. Free energy and chemical equilibrium
- 6. Temperature and pressure dependence of phase equilibrium
- 7. Thermodynamics of solutions
- 8. Colligative properties of solutions
- 9. Phase and surface properties
- 10. Electrolytes in solutions
- 11. Electrochemical cells
- 12. Kinetics of chemical reactions
- 13. Reaction rate and mechanism
- 14. Spectroscopies and diffractions
- 15. Summary

	[Title]			[Instructor]		
Advanced Inorganic Chemistry		Satoshi Wada / Hiroshi Yanagi		shi Yanagi		
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]	
GTG532	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester	Thu./I	English/ Japanese	

Students learn the basics and application of electron behavior in solids

[Objectives]

To understand fundamental principle of electronic and optical properties of solids

[Requirements]

A good grounding in Physical Chemistry, Inorganic Chemistry, and Quantum Chemistry.

[Evaluation]

- 1 Midterm examination 30%
- 2 homework 30%
- 3 class participation 40%

[Textbooks]

[References]

- *The lectures from 1 to 5 are held with Zoom or Teams.
- *The lectures from 6 to 10 are held on demand.
- *The lectures from 10 to 15 are held with Zoom.
- 1. Introduction
- 2. Crystal Structure
- 3. Chemical bonding and band structure
- 4. Spectroscopic methods
- 5. Material design based on electronic structure
- 6. The essence of electronic structure
- 7. Material design based on electronic structure
- 8. Mechanism of electric polarization
- 9. Complex dielectric constant and dielectric relaxation
- 10. Evaluation of dielectric properties
- 11. Electrical conductivity
- 12. Defect and nonstoichiometry in solid
- 13. Mechanism of electronic conductivity
- 14. Mechanism of ionic conductivity
- 15. Evaluation of electrical conductivity

	[Title]			[Instructor	
Advanced Inorganic Chemistry		Satoshi Wada / Hiroshi Yanagi		shi Yanagi	
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG532	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester Thu./I		English/ Japanese

Students learn the basics and application of electron behavior in solids

[Objectives]

To understand fundamental principle of electronic and optical properties of solids

[Requirements]

A good grounding in Physical Chemistry, Inorganic Chemistry, and Quantum Chemistry.

[Evaluation]

- 1 Midterm examination 30%
- 2 homework 30%
- 3 class participation 40%

[Textbooks]

[References]

- 1. Introduction
- 2. Crystal Structure
- 3. Chemical bonding and band structure
- 4. Spectroscopic methods
- 5. Other evaluation method
- 6. The essence of electronic structure
- 7. Material design based on electronic structure
- 8. Midterm examination
- 9. Mechanism of electric polarization
- 10. Complex dielectric constant and dielectric relaxation
- 11. Evaluation of dielectric properties
- 12. Ferroelectrics and ferroelectric domain configuration
- 13. Piezoelectricity
- 14. Application of dielectrics and ferroelectrics
- 15. Summative assessment for total score

[Title]		[Instructor]			
Advanced Materials Chemistry			aka / Takahi Eiichi Kondo		
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG533	2	Special Educational Program for Green Energy Conversion Science and Technology	2st Semester Tue./I		English/ Japanese

Synthesis and crystal growth techniques for solid state materials are acquired on the base of phase equilibrium in this course. Also crystal chemistry and characterization for solid state materials are acquired. For various synthesis processes, the mechanism and their theories are acquired.

[Objectives]

- 1. to understand relationship between defect concentration and physical properties by lattice defect formation in crystalline materials
- 2. to understand formation mechanism in various synthesis processes for solid state materials
- 3. to gain ability to use binary phase diagrams

[Requirements]

inorganic chemistry, solid state chemistry, materials engineering, physical chemistry, electronic physical properties

[Evaluation]

homework/ examination: 70%

audit attitude : 10% presentation : 20%

[Textbooks]

Anthony R. West, Solid State Chemistry and Its Applications, Second Edition, JOHN WILEY & SONS, LTD, ISBN:978-1-119-94294-8

[References]

- 1. Function and property by crystal defects
- 2. Non-stoichiometry and lattice defects in oxides
- 3. Defect concentration and defect equilibrium
- 4. Relationship between defect concentration and electrical conductivity
- 5. Interim summary I
- 6. Synthesis of inorganic materials by solid state reaction
- 7. Sol-gel synthesis of inorganic materials
- 8. Synthesis by hydrothermal and solvothermal reactions
- 9. Thin film preparation by gas phase reaction
- 10. Solid-liquid interface & Interim summary II
- 11. Phase rule and phase diagram
- 12. Phase diagram and microstructures
- 13. Solid-liquid interface and its equilibrium
- 14. Basic theories of solution chemistry and phase diagram
- 15. Nucelation and crystal growth Summary

[Title]		[Instructor]			
Advanced Course of Materials Design for Fuel Cells			vatake / Akih ohara / Jung		
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG534	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester Tue./II		English/ Japanese

Fuel cells are electric power supply devices, which convert chemical energy to electric energy directly and reciprocally. Among them, polymer electrolyte fuel cells (PEFCs) for electric vehicles, portable devices, and residential power supply and solid oxide fuel cells (SOFCs) as on-site power generation have attracted a considerable attention. In this class, principle, design and evaluation of these fuel cells and their component materials will be discussed.

[Objectives]

To understand principle and evaluation of PEFCs and SOFCs and their component materials

[Requirements]

Basic knowledge on electrochemistry and physical chemistry

[Evaluation]

Report and examination: 50%

Mark given for class participation: 50%

[Textbooks]

None

[References]

Denkikagakugairon (co-authored by Matsuda and Iwakura), Maruzen, ISBN: 4621039962

- 1. Electrochemistry of fuel cells 1
- 2. Electrochemistry of fuel cells 2
- 3. Principle and research trend of fuel cells 1
- 4. Principle and research trend of fuel cells 2
- 5. Design of fuel cell electrocatalysts: cathode catalysts 1
- 6. Design of fuel cell electrocatalysts: cathode catalysts 2
- 7. Design of fuel cell electrocatalysts: anode catalysts 1
- 8. Design of fuel cell electrocatalysts: anode catalysts 2
- Methanol oxidation catalysts 1
- 10. Methanol oxidation catalysts 2
- 11. Design of highly dispersed catalysts 1
- 12. Design of highly dispersed catalysts 2
- 13. Design of functional materials 1
- 14. Design of functional materials 2
- 15. Summary

	[Title]			[Instructor]			
Advanced Course of Engineering for Solar Energy Conversion			Irie / Hirosl hihiroTakas	~			
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]		
GTG535	2	Special Educational Program for Green Energy Conversion Science and Technology	1st Semester Mon./II		English/ Japanese		

A light-related system is one of the candidate technologies for sustainable energy conversion and environmental preservation. We will learn such light-related systems based on mainly physical chemistry as well as quantum chemistry and solid state physics. Students also learn the fundamental principle of standard and new concept solar cells.

[Objectives]

- 1. To understand the interaction of light with solids, and successive phenomena
- 2. To understand the fundamental principle of standard and new concept solar cells

[Requirements]

Physical Chemistry, Quantum chemistry, Solid state physics, Inorganic Chemistry, and Semiconductor Physics

[Evaluation]

- 1 final examination 20%
- 2 midterm examination 20%
- 3 homework / reports 20%
- 4 class participation / presentation 40%

[Textbooks]

[References]

魚崎浩平、米田龍、高橋誠、金子晋(共訳): 固体の電子構造と化学、技報堂出版、1989 年(in Japanese) 山口 真史・M·A·グリーン・大下 祥雄・小島 信晃, 太陽電池の基礎と応用, 丸善 (in Japanese)

Martin A. Green, Solar Cells, University of New South Wales

Peter Wurfel, 太陽電池の物理, 丸善 (in Japanese)

Peter Würfel, Physics of Solar Cells: From Basic Principles to Advanced Concepts, Wiley-VCH

- 1.Introduction
- 2. Light energy conversion, Basic theory
- 3. Solar energy conversion: To chemical energy 1
- 4. Solar energy conversion: To chemical energy 2
- 5 Solar energy conversion : To hydrogen energy
- 6. Thermal energy conversion: Basic theory
- 7. Thermal energy conversion: To electricity
- 8. Solar cells and sunlight
- 9. Semiconductor properties
- 10. Carrier generation and recombination
- 11. Si based solar cells
- 12. Compound-semiconductor Solar cells
- 13. Organic solar cells
- 14. Future view
- 15. Final examination / presentation

[Title]				[Instructors]
Advanced Course of Science for Surfaces and Interfaces			kai, Akiyosh oshihiro Miy		
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG536	2	Special Educational Program for Green Energy Conversion Science and Technology	1st Fri /I		English/ Japanese

Comprehending basic surface crystallography, surface analytical methods, and surface reactions to be applied to students' researches.

[Objectives]

Understanding basic ideas of the following topics:

- 1) Surface and interface science.
- 2) Surface crystallography at the atomic level.
- 3) Surface analytical methods.
- 4) Surface reactions on model and real surfaces.
- 5) Adsorption and reactions on solid surfaces.

[Requirements]

Basic knowledge on solid crystallography and quantum chemistry.

[Evaluation]

Class participation 40%

Reports, quiz, and examination 60%

[Textbooks]

[References]

Atkins' Physical Chemistry, Peter Atkins.

- 1. Introduction
 - Surface crystallography I: Single crystal surfaces
- 2. Surface crystallography II: Notification of surface structures; adlayers
- 3. Surface crystallography III: Reciprocal lattice
- 4. Surface crystallography IV: Reciprocal lattice to real lattice
- 5. Quiz on surface crystallography
- 6. Surface analysis method I: Electrochemistry on Single crystal surfaces
- 7. Surface analysis method II: Morphological study with Scanning Probe Microscopy
- 8. Surface analysis method III: Infrared absorption spectroscopy
- 9. Surface analysis method IV: Surface enhanced Raman spectroscopy
- 10. Quiz on surface analysis method
- 11. Adsorption at solid surfaces I: (interpretation of adsorption isotherms)
- 12. Adsorption at solid surfaces II: (surface characterization using adsorption phenomena)
- 13. Adsorption at solid surfaces III: (chemisorption and surface catalysis)
- 14. Adsorption at solid surfaces IV: (catalytic reaction mechanisms at solid surfaces)
- 15. Quiz on adsorption at solid surfaces

	[Title]			[Instructor]	
Advanced Course of Polymer Material Chemistry		Hidenori Okuzaki		aki	
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]
GTG537	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester	Thu./II	English/ Japanese

Polymer is one of three major materials together with metals and ceramics, which has been widely utilized in the fields of aerospace, electronics, telecommunication, transportation, and medicines. In this class, basic and advanced technologies in structure and function of various functional polymer materials such as gels, liquid crystals, conducting polymers, piezoelectric polymers, and intelligent materials will be discussed.

[Objectives]

To understand advanced technology in structure and function of various functional polymer materials.

[Requirements]

Basic knowledge on organic chemistry, physical chemistry, and polymer chemistry.

[Evaluation]

Mark given for class participation and report: 50%

Presentation and examination: 50%

[Textbooks]

None

[References]

Zukai Kobunshi zairyou saizensen (Kunihiro Ozaki and Kazuo Matsuura), Kogyotyosakai, ISBN:476934161X

- 1. Polymer materials chemistry (introduction)
- 2. Research and development of polymer materials
- 3. Structure of polymer materials (molecular weight and distribution)
- 4. Structure of polymer materials (tacticity)
- 5. Structure of polymer materials (crystallinity, crystallite size, and crystalline orientation)
- 6. Structure of polymer materials (cross-linking and gels)
- 7. Function of polymer materials (high modulus and high strength polymers)
- 8. Function of polymer materials (biocompatibility and medical polymers)
- 9. Function of polymer materials (polymer gels)
- 10. Function of polymer materials (semiconducting polymers)
- 11. Function of polymer materials (conducting polymers)
- 12. Function of polymer materials (plastic electronics)
- 13. Function of polymer materials (intelligent polymer materials)
- 14. Function of polymer materials (biomimetic polymers)
- 15. Summary

	[Title]			[Instructor]			
Advanced	Course of F	ourse of English for Green Energy Science and Technology, Elementary Level D. A. Tryk / M. E.		. Brito			
[Code]	[Credits]	[Program]	[Semester]	[Hours]	[Language of instruction]		
GTG538	2	Special Educational Program for Green Energy Conversion Science and Technology	1st Semester	Mon./I	English/ Japanese		

This course will cover reading, writing, speaking and listening for scientific and engineering English, including special emphasis on green energy and technology. Presentation and discussion skills will be particularly emphasized.

[Objectives]

The specific achievements or milestones will include: (1) ability to read a technical paper and summarize it briefly in English; (2) ability to write a short paper; (3) ability to confidently give a short technical presentation in English; (4) ability to ask questions at an oral technical presentation.

[Requirements]

M1 status

[Evaluation]

Attendance: 20%; presentations: 40%; reports: 40%

[Textbooks]

None

[References]

None

- 1. Introduction; overview; basic pronunciation;
- 2. Pronunciation of general chemical terms, specific terms for student research themes; self-introductions;
- 3. Brief self-introductions
- 4. Short, informal oral presentations 1
- 5. Short, informal oral presentations 2
- 6. Short, informal oral presentations 3
- 7. Brief oral introduction to your research field for non-specialists 1
- 8. Brief oral introduction to your research field for non-specialists 2
- 9. Brief oral introduction to your research field for non-specialists 3
- 10. Brief oral introduction to your research field for non-specialists 4
- 11. Final oral presentations 1
- 12. Final oral presentations 2
- 13. Final oral presentations 3
- 14. Final oral presentations 4
- 15. Final oral presentations 5

[Title]				[Instructor]		
Exercises for Green Energy Conversion IA		all academic supervisors		rvisors		
[Code]	[Credits]	[Program]	Samagtari Halirgi		[Language of instruction]	
GTG612	1	Special Educational Program for Green Energy Conversion Science and Technology	1nd Semester	/	English/ Japanese	
[Outline ar	nd purpose]		1			
The purpos	se is to acqu	ire knowledge about the green energy conversion s	cience and tec	chnology.		
[Objectives]					
Students v	will have v	will have acquired the ability to address follow	ing issues th	nrough our	instructional	

Students will have will have acquired the ability to address following issues through our instructional approach, which integrates basic and practical studies; Fuel cells, Solar energy conversion, Energy-conversion materials, New energy technology

[Requirements]

Physical chemistry, Materials Physics and Chemistry, Catalyst Science, Environmental Science, etc.

[Evaluation]

Laboratory work, Technical report

[Textbooks]

[References]

- ·Students have opportunities to present the results of their studies in monthly research meetings.
- Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title]				[Instructor]		
	Exercis	es for Green Energy Conversion IB	all aca	ademic supe	rvisors	
[Code]	[Credits]	[Program]	[Semester]	Semester] [Hours] [Language instruction		
GTG613	1	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester	/	English/ Japanese	
[Outline ar	nd purpose]				•	
The purpos	se is to acqu	ire knowledge about the green energy conversion s	cience and tec	hnology.		
[Objectives]					
		vill have acquired the ability to address follow rates basic and practical studies; Fuel cells, Solar	-	_		

[Requirements]

Physical chemistry, Materials Physics and Chemistry, Catalyst Science, Environmental Science, etc.

[Evaluation]

Laboratory work, Technical report

materials, New energy technology

[Textbooks]

[References]

- ·Students have opportunities to present the results of their studies in monthly research meetings.
- Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title]		[Instructor]				
Exercises for Green Energy Conversion IIA		all academic supervisors		rvisors		
[Code]	[Credits]	[Program]	[Semester]	[Semester] [Hours] [Language instruction		
GTG614	1	Special Educational Program for Green Energy Conversion Science and Technology	1nd Semester	/	English/ Japanese	
Outline ar	nd purpose]					
The purpos	se is to acqu	ire knowledge about the green energy conversion s	cience and tec	hnology.		
[Objectives	1					

Students will have will have acquired the ability to address following issues through our instructional approach, which integrates basic and practical studies; Fuel cells, Solar energy conversion, Energy-conversion materials, New energy technology

[Requirements]

Physical chemistry, Materials Physics and Chemistry, Catalyst Science, Environmental Science, etc.

[Evaluation]

Laboratory work, Technical report

[Textbooks]

[References]

- ·Students have opportunities to present the results of their studies in monthly research meetings.
- Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title] Exercises for Green Energy Conversion IIB		[Instructor] all academic supervisors			
					[Code]
GTG615	1	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester	/	English/ Japanese
[Outline ar	nd purpose]				•
		ire knowledge about the green energy conversion s	cience and tec	hnology.	
[Objectives	<u>s]</u>				
approach,	which integ	will have acquired the ability to address follow trates basic and practical studies; Fuel cells, Solar technology	_	~	

[Requirements]

Physical chemistry, Materials Physics and Chemistry, Catalyst Science, Environmental Science, etc.

[Evaluation]

Laboratory work, Technical report

[Textbooks]

[References]

- ·Students have opportunities to present the results of their studies in monthly research meetings.
- Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- \cdot The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title] Professional Research for Green Energy Conversion IA		[Instructor] all academic supervisors			
					[Code]
GTG616	2	Special Educational Program for Green Energy Conversion Science and Technology	1st Semester	/	English/ Japanese
[Objectives]	1				
Students w Develop in	ill have acc nnovative (low-carbon	omplished the following general and specific learning technology in order to efficiently and economical, sustainable society. nce of various energy conversion devices and through	lly convert an		een energy to
[Requireme Physical ch		aterials Physics and Chemistry, Catalyst Science, I	Environmental	l Science, et	c.

[Textbooks]

[References]

- \cdot Students are required to research green energy conversion-related issues in each laboratory.
- \cdot Students have opportunities to present the results of their studies in monthly research meetings.
- · Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title] Professional Research for Green Energy Conversion IB		[Instructor] all academic supervisors			
					[Code]
GTG617	2	Special Educational Program for Green Energy Conversion Science and Technology	2nd Semester	/	English/ Japanese
[Outline ar	nd purpose]				
		ire knowledge about the green energy conversion s	cience and tec	hnology.	
Objectives		1.1 1.1 6.11			
·Develop i establish a	nnovative t low-carbon	complished the following general and specific learning technology in order to efficiently and economical sustainable society. Ince of various energy conversion devices and through the contraction of the	lly convert an		een energy to
[Requireme			8		
		aterials Physics and Chemistry, Catalyst Science, I	Environmental	l Science, et	c.
[Evaluation	n]				
T 1 /	1 77 1	nical report			

Laboratory work, Technical report

[Textbooks]

[References]

- \cdot Students are required to research green energy conversion-related issues in each laboratory.
- ·Students have opportunities to present the results of their studies in monthly research meetings.
- · Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- ·The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Title] Professional Research for Green Energy Conversion IIA		[Instructor] all academic supervisors			
					[Code]
GTG618	2	Special Educational Program for Green Energy Conversion Science and Technology	1st Semester	/	English/ Japanese
[Outline ar	nd purpose]				1
The purpos	se is to acqu	ire knowledge about the green energy conversion s	cience and tec	hnology.	
[Objectives	<u>s]</u>				
		omplished the following general and specific learns			
		echnology in order to efficiently and economical	lly convert an	nd store gr	een energy to
		, sustainable society. nce of various energy conversion devices and throu	gh thoso studi	ios	
Requirem		nce of various energy conversion devices and throu	ign these studi	ies	
		aterials Physics and Chemistry, Catalyst Science, l	Environmental	Science, et	tc.
Physical cl	-				
Physical cl					
Physical cl					
Physical cl	n]				
[Evaluatio		nical report			
[Evaluatio		nical report			
[Evaluatio Laboratory	work, Tech	nical report			
[Evaluatio	work, Tech	nical report			
[Evaluatio Laboratory	work, Tech	nical report			

- $\cdot \text{Students are required to research green energy conversion-related issues in each laboratory}.$
- \cdot Students have opportunities to present the results of their studies in monthly research meetings.
- · Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.

[Code] [Credits]	ch for Green Energy Conversion IIB [Program]	all aca	ademic supe		
Snee	[Program]		all academic supervisors		
GTG619 2 Spec		[Semester]	[Hours]	[Language of instruction]	
2	cial Educational Program for Green Energy Conversion Science and Technology	2nd Semester	/	English/ Japanese	
	nowledge about the green energy conversion s	erence and tee	miorogy.		
·Develop innovative techno establish a low-carbon, sust	ished the following general and specific learning ology in order to efficiently and economical ainable society. If various energy conversion devices and through	lly convert ar	nd store gro	een energy to	
[Requirements] Physical chemistry, Materia	lls Physics and Chemistry, Catalyst Science, I	Environmenta	l Science, et	c.	

$[{\bf References}]$

[Textbooks]

- \cdot Students are required to research green energy conversion-related issues in each laboratory.
- ·Students have opportunities to present the results of their studies in monthly research meetings.
- · Subjects of interactive discussion between students and faculty are provided. A particular focus is on developing debate skills in English through interactive discussion subjects presented by foreign faculty.
- •The number of faculty is large enough to maintain a student to faculty ratio of 1.5 to 1, creating small-group instruction, with close attention to each student.